5N. 09/765-925 to overcome 21 Jan

=> dis 18 1- sub bib abs
YOU HAVE REQUESTED DATA FROM 3 ANSWERS - CONTINUE? Y/(N):y

L8 ANSWER 1 OF 3 REGISTRY COPYRIGHT 2003 ACS

RN 125529-39-7 REGISTRY

CN 1,2-Cyclobutanedicarboxylic acid, 3,4-bis(4-hydroxyphenyl)-, (1R,2S,3R,4S)-rel- (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

FS STEREOSEARCH

MF C18 H16 O6

SR CA

LC STN Files: CA, CAPLUS, TOXCENTER

Relative stereochemistry.

PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

2 REFERENCES IN FILE CA (1962 TO DATE)

2 REFERENCES IN FILE CAPLUS (1962 TO DATE)

REFERENCE 1

AN 135:117221 CA

TI Diphenylcyclobutane dicarboxylate derivatives as analgesics and antiinflammatory agents

IN Nakamura, Motoyuki; Yoshizawa, Toyokichi; Chitama, Akira; Nohara, Toshihiro; Sakurada, Shinobu

PA Showa Pharmaceutical Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

LAM.		Т																
	PA	rent	NO.		KII	ND :	DATE			A1	PPLI	CATI	ON NC	ο.	DATE			
							7 – – –											
PI	JP	2001	1998	84	A.	2 /	2001	0724		J	20	00-1	3449		2000	0121		
	ΕP	1118	324		A.	1 :	2001	0725		El	200	01-1	0121	3 .	2001	0119		
		R:	AT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	IT,	LI,	LU,	NL,	SE,	MC,	PT,
			ΙE,	SI,	LT,	LV,	FΙ,	RO										
PRAI	JΡ	2000	-134	49	20	0001	21											

GI

AΒ Diphenylcyclobutane dicarboxylate derivs. (I; X1, X2, Y1, Y2, Z1, Z2 = H, OH, halogen, alkyl, alkoxy, N; R1, R2 = OH, halogen, alkoxyl, aryloxy, turpenoxy, sugar, N) are claimed as analgesics and antiinflammatory agents with min. toxicity. I were prepd., and their analgesic and antiinflammatory actions and effect on gastric mucosa were tested.

REFERENCE 2

AN112:115748 CA

TICyclodimers of p-coumaric and ferulic acids in the cell walls of tropical grasses

ΑU Ford, Clive W.; Hartley, Roy D.

CS AFRC Inst. Grassl. Anim. Prod., Hurley/Maidenhead/Berkshire, SL6 5LR, UK

SO Journal of the Science of Food and Agriculture (1990), 50(1), 29-43 CODEN: JSFAAE; ISSN: 0022-5142

DTJournal

LΑ English

AΒ

Cell walls from leaves and stems of three tropical grasses (Setaria anceps cv Nandi), pangola grass Digitara decumbens, and spear grass Heteropogon contortus were extd. with alkali, and ether-sol. fractions were prepd. from the acidified solns. Cyclobutane dimers derived from p-coumaric acid (CA: 3-(4-hydroxphenyl) propenoic acid) and ferulic acid (FA: 3-(4-hydroxy-3-methoxyphenyl) propenoic acid) were found in all plant residues. Structural information on the dimers was obtained by gas chromatog.-mass spectrometry. Stem cell wall differed from those of leaves in having major dimer components which contained residues tentatively identified as coniferyl alc. (3-(4-hydroxy-3-methoxyphenyl)-2propen-1-ol). Species differences in leaf cell wall cyclodimers were evident. Pangola and setaria grasses had dimers mainly derived from FA, whereas spear grass had over 70% of dimers derived from CA. Di-esterification of some dimers in the cell walls was confirmed by the release of dimers contg. the reduced forms of CA and FA (coumaryl and coniferyl alcs. resp.) when leaf cell walls were extd. with borohydride. Heat to tail dimerization giving derivs. of truxillic acid (t-2,c-4-diphenyl-r-1,t-3-cyclobutanedicarboxylic acid) predominated, although evidence was obtained for the presence of head-to-head coumaric acid dimers (derivs. of truxinic acid, t-3,t-4-diphenyl-r-1,c-2cyclobutanedicarboxylic acid) in setaria stem and spear grass leaf. results suggested that cyclodimers in grass cell walls occurred in crosslinking structures of varying complexity between macromols., where they would possibly contribute to fiber strength and thus have an adverse effect on the nutritional value of forage.

- L8 ANSWER 2 OF 3 REGISTRY COPYRIGHT 2003 ACS
- RN 104496-04-0 REGISTRY
- 1,2-Cyclobutanedicarboxylic acid, 3,4-diphenyl-, bis(4-nitrophenyl) ester, CN (1R, 2S, 3R, 4S) - rel - (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

1,2-Cyclobutanedicarboxylic acid, 3,4-diphenyl-, bis(4-nitrophenyl) ester, (1.alpha., 2.alpha., 3.beta., 4.beta.) -OTHER NAMES:

CN .beta.-Truxinic acid bis(4-nitrophenyl) ester

CN Bis(p-nitrophenyl) .beta.-truxinate

FS STEREOSEARCH

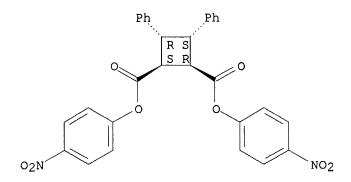
MF C30 H22 N2 O8

CI COM

SR CA

LC STN Files: CA, CAPLUS, CASREACT, TOXCENTER

Relative stereochemistry.



PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

4 REFERENCES IN FILE CA (1962 TO DATE)

4 REFERENCES IN FILE CAPLUS (1962 TO DATE)

REFERENCE 1

AN 135:117221 CA

TI Diphenylcyclobutane dicarboxylate derivatives as analgesics and antiinflammatory agents

IN Nakamura, Motoyuki; Yoshizawa, Toyokichi; Chitama, Akira; Nohara, Toshihiro; Sakurada, Shinobu

PA Showa Pharmaceutical Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 8 pp.

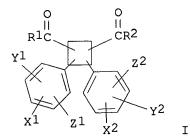
CODEN: JKXXAF

DT Patent

LA Japanese

FAN. CNT 1

FAN.CNT I																		
	PAT	CENT	NO.		KI	ND	DATE			A.	PPLI	CATI	ON No	Э.	DATE			
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PI	JP	2001	1998	84	A.	2	2001	0724		J	P 20	00-1	3449		2000	0121		
	ΕP	1118	324		A.	1	2001	0725		E	P 20	01-1	0121	3	2001	0119		
		R:	AT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	IT,	LI,	LU,	NL,	SE,	MC,	PT,
			ΙE,	SI,	LT,	LV,	FI,	RO										
PRAI	JP	2000	-134	49	20	0001	21											
GI												•						



AΒ

OH, halogen, alkyl, alkoxy, N; R1, R2 = OH, halogen, alkoxyl, aryloxy, turpenoxy, sugar, N) are claimed as analgesics and antiinflammatory agents with min. toxicity. I were prepd., and their analgesic and antiinflammatory actions and effect on gastric mucosa were tested.

REFERENCE 2

AN 110:76172 CA

TI Polycondensation of bis(p-nitrophenyl) .beta.-truxinate with diamine

AU Iizawa, Takashi; Miyaji, Takatoshi; Nishikubo, Tadatomi

CS Fac. Eng., Kanagawa Univ., Yokohama, 221, Japan

SO Journal of Polymer Science, Part A: Polymer Chemistry (1988), 26(11), 2941-52

CODEN: JPACEC; ISSN: 0887-624X

DT Journal

LA English

Model reaction of bis(4-nitrophenyl) .beta.-truxinate (I) with aliph. AΒ amines proceeded quant. at room temp. Accordingly, polycondensation of I with various diamines was carried out at room temp. or 80.degree.. During the polycondensation with diamines, the pptn. of polymer or gelation of the polymn. soln. occurred, which could limit the mol. wt. of the polymer. On the other hand, the reaction of I with 1,3-bis(4-piperidy1)propane proceeded homogeneously to give a copolymer (II) with relatively high mol. wt., and II showed excellent soly. in many solvents. TG and DTA indicated that the obtained polymers were more stable at lower temps. than around 270.degree.. The polymer prepd. from the polycondensation of I with hexamethylenediamine showed m.p. and decompn. due to imidation at 282.degree.. Photochem. reactions of these polymers were carried out in the film state. Irradn. with 254 nm light caused an absorption at 272 nm to appear and the mol. wt. to decrease. This meant that the scission of cyclobutane rings in the main chain occurred to give cinnamamide structures. The absorption at 272 nm was decreased by irradn. with 302.5 nm light. The UV spectrum of the irradiated polymer differed from that of the original polymer. These results suggested that dimerization of the resulting cinnamamide moieties occurred in competition with their trans-cis-isomerization. On the other hand, the rate of scission of cyclobutane rings of II was faster than that of the corresponding polyamide contg. an .alpha.-truxillamide structure.

REFERENCE 3

AN 106:66920 CA

TI Cyclobutane compounds

IN Nishikubo, Tatatomi

PA Research Development Corp. of Japan, Japan

SO Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF

DT Patent

10 6

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE		
PI	JP 61212532	A2	19860920	JP 1985-53392	19850319		
	JP 05088210	B4	19931221				
PRAI GI	JP 1985-53392	19850	319				

AΒ

as optical materials, are prepd. by photocycloaddn. of PhCH:CHCO2R (II) in the presence of a dispersant solvent. Thus, 0.059 mol II (R = p-O2NC6H4) in a mixed solvent of 380 mL and 20 mL hexane-MeOH at a dispersion of 20 g/L was irradiated 6 h at 30.degree. with stirring to give 91.2% I (R = p-O2NC6H4).

REFERENCE 4

AN 105:152559 CA

TI Convenient synthesis of .beta.-truxinic acid via photodimerization of p-nitrophenyl cinnamate in the crystalline state

AU Nishikubo, Tadatomi; Takahashi, Eiji; Miyaji, Takatoshi; Iizawa, Takashi

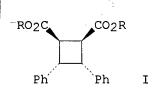
CS Fac. Eng., Kanagawa Univ., Yokohama, 221, Japan

SO Bulletin of the Chemical Society of Japan (1985), 58(11), 3399-400 CODEN: BCSJA8; ISSN: 0009-2673

DT Journal

LA English

GI



AB .beta.-Truxinic acid (I, R = H) was prepd. in high yield via topochem. photodimerization of PhCH:CHCO2C6H4NO2-4 (II) crystals, followed by hydrolysis of the obtained dimer. Thus, II crystals were suspended in hexane under N and irradiated with a 100-W high pressure Hg lamp to give 99.2% I (R = 4-O2NC6H4) which was treated with KOH in MeOH and then concd. HCl to give 96.7% I (R = H).

L8 ANSWER 3 OF 3 REGISTRY COPYRIGHT 2003 ACS

RN 528-34-7 REGISTRY

CN 1,2-Cyclobutanedicarboxylic acid, 3,4-diphenyl-, (1R,2S,3R,4S)-rel- (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN 1,2-Cyclobutanedicarboxylic acid, 3,4-diphenyl-,

(1.alpha., 2.alpha., 3.beta., 4.beta.) -

CN 1,2-Cyclobutanedicarboxylic acid, 3,4-diphenyl-, cis-1,2,trans-1,3,trans-1,4- (8CI)

OTHER NAMES:

CN .beta.-Truxinic acid

CN Isotruxillic acid

FS STEREOSEARCH

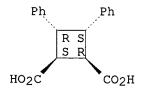
MF C18 H16 O4

CI COM

LC STN Files: AGRICOLA, BEILSTEIN*, BIOSIS, CA, CAPLUS, CASREACT, CHEMCATS, TOXCENTER

(*File contains numerically searchable property data)

Relative stereochemistry.



^{**}PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT**

- 36 REFERENCES IN FILE CA (1962 TO DATE)
- 1 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA
- 36 REFERENCES IN FILE CAPLUS (1962 TO DATE)

REFERENCE 1

- AN 138:39088 CA
- TI Topochemical reaction control in solution. III. 4,15-Diamino[2.2]paracyclophane, a reusable template for topochemical reaction control in solution
- AU Zitt, Holger; Dix, Ina; Hopf, Henning; Jones, Peter G.
- CS Institut fur Organische Chemie, Technische Universitat Braunschweig, Braunschweig, 38106, Germany
- SO European Journal of Organic Chemistry (2002), (14), 2298-2307 CODEN: EJOCFK; ISSN: 1434-193X
- PB Wiley-VCH Verlag GmbH
- DT Journal
- LA English
- An efficient synthesis of [2.2]paracyclophane-4,15-dicarboxylic acid from [2.2]paracyclophane (I) has been developed. The diacid was converted via a diazide into the 4,15-diisocyanato[2.2]paracyclophane (II), a versatile intermediate that could be transformed into many new pseudogeminally substituted derivs. of I. For example, treatment of II with alcs. provided two carbamates. On treatment of II with diisopropylamine, a urea compd., was obtained, whereas redn. with lithium aluminum hydride afforded a cyclic urea (III). Hydrolysis of II furnished a diamine (IV), which was used as a reusable spacer in a [2+2]photoaddn. expt. Thus, treatment of IV with trans-cinnamoyl chloride provided a bis(amide), which on irradn. in acetone ring-closed to give a cyclobutane compd. Sapon. of this yielded 3,4-diphenyl-1,2-cyclobutanedicarboxylic acid (.beta.-truxinic acid) and returned the spacer system 19, both in quant. yield. The X-ray structures of II and III are reported.
- RE.CNT 37 THERE ARE 37 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

REFERENCE 2

- AN 135:117221 CA
- TI Diphenylcyclobutane dicarboxylate derivatives as analgesics and antiinflammatory agents
- IN Nakamura, Motoyuki; Yoshizawa, Toyokichi; Chitama, Akira; Nohara, Toshihiro; Sakurada, Shinobu
- PA Showa Pharmaceutical Co., Ltd., Japan
- SO Jpn. Kokai Tokkyo Koho, 8 pp. CODEN: JKXXAF
- DT Patent
- LA Japanese
- FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
		-			
ΡI	JP 200119988	4 A2	20010724	JP 2000-13449	20000121
	EP 1118324	A1	20010725	EP 2001-101213	20010119
	R: AT,	BE, CH, D	E, DK, ES, FR,	GB, GR, IT, LI, LU	, NL, SE, MC, PT,
	IE,	SI, LT, L	V, FI, RO		

PRAI JP 2000-13449 20000121

GI

Diphenylcyclobutane dicarboxylate derivs. (I; X1, X2, Y1, Y2, Z1, Z2 = H, OH, halogen, alkyl, alkoxy, N; R1, R2 = OH, halogen, alkoxyl, aryloxy, turpenoxy, sugar, N) are claimed as analgesics and antiinflammatory agents with min. toxicity. I were prepd., and their analgesic and antiinflammatory actions and effect on gastric mucosa were tested.

REFERENCE 3

AN 133:321558 CA

TI Compelled Orientational Control of the Solid-State Photodimerization of trans-Cinnamamides: Dicarboxylic Acid as a Non-covalent Linker

AU Ito, Y.; Hosomi, H.; Ohba, S.

CS Graduate School of Engineering, Department of Synthetic Chemistry and Biological Chemistry, Kyoto University, Kyoto, 606-8501, Japan

SO Tetrahedron (2000), 56(36), 6833-6844

Ι

CODEN: TETRAB; ISSN: 0040-4020

PB Elsevier Science Ltd.

DT Journal

LA English

The 2:1 hydrogen-bonded cocrystals la.cntdot.ox, la.cntdot.su, AB la.cntdot.pht, la.cntdot.fu, lb.cntdot.ox, lc.cntdot.ox, ld.cntdot.ox between 4-R-trans-cinnamamides [1a-11c: R = H, Me, Cl, resp.; 1d = trans-3-(2-thienyl)acrylamide] and dicarboxylic acids (ox, su, gl, fu, pht) were prepd. and characterized by IR and powder X-ray techniques. crystal structures of la.cntdot.pht, la.cntdot.ox and la.cntdot.fu were solved by single crystal X-ray diffraction. Phthalic acid (pht) caused .beta.-type photodimerization of trans-cinnamamide (1a) in the cocrystal and functioned as a non-covalent linker like gauche 1,2-diamines in photodimerization of trans-cinnamic acids. Oxalic acid (ox) forced la to take a bilayer structure that is suitable for .beta.-type photodimerization. In the case of fumaric acid (fu), cross photodimerization with la occurred to give a cycloadduct. For the cocrystals la.cntdot.pht and la.cntdot.fu, pedal-like motion was assumed to occur prior to the dimerization.

RE.CNT 37 THERE ARE 37 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

REFERENCE 4

AN 129:148662 CA

- TI A controlled photodimerization of cinnamic acid in a composite bilayer of dioctadecyldimethylammonium bromide
- AU Nakamura, Tomotaka; Takagi, Katsuhiko; Sawaki, Yasuhiko
- CS Department of Applied Chemistry, Graduate School of Engineering, Nagoya University, Nagoya, 464-01, Japan
- SO Molecular Crystals and Liquid Crystals Science and Technology, Section A: Molecular Crystals and Liquid Crystals (1998), 313, 341-346 CODEN: MCLCE9; ISSN: 1058-725X
- PB Gordon & Breach Science Publishers
- DT Journal
- LA English
- AB Investigations of the photochem. of cinnamic acid (CA) were carried out in a form of 1: 1 composite solids with dioctadecyldimethylammonium bromide (DODAB) or chloride (DODAC). UV irradn. of the composite solids with DODAB resulted in the highly selective formation of the syn-Head-to-Head

dimer, suggesting the parallel orientation of CA within the complex. Studies of X-ray diffraction, DSC, and IR spectra revealed the formation of a composite bilayer assembly showing the interaction between polar groups of CA and DODAB as well as the incorporation of CA into the alkyl regions of DODAB. On the other hand, the mixt. of CA and DODAC showed less and non-selective photodimerization reaction, which could be ascribed to a disordered mol. arrangement, judging from a weak and broad endothermic peak at around reaction temp. in the DSC curve.

RE.CNT 24 THERE ARE 24 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

REFERENCE 5

- AN 129:15910 CA
- TI Photodimerization of trans-cinnamic acid in a bilayer of dimethyldioctadecylammonium bromide
- AU Nakamura, Tomotaka; Takagi, Katsuhiko; Sawaki, Yasuhiko
- CS Dep. Applied Chemistry, Fac. Eng., Nagoya Univ., Nagoya, 464-01, Japan
- SO Bulletin of the Chemical Society of Japan (1998), 71(4), 909-914 CODEN: BCSJA8; ISSN: 0009-2673
- PB Chemical Society of Japan
- DT Journal
- LA English
- AB The photochem. of trans-cinnamic acid (trans-1) was studied in a cast film of dimethyldioctadecylammonium bromide (2), deposited from aq. dispersions as a cryst. solid by the spontaneous vaporization of water. UV irradn. of the cast film resulted in a highly selective formation of the syn-Head-to-Head dimer, suggesting a parallel orientation of trans-1 within a bilayer film. Analyses with X-ray diffraction, differential scanning calorimetry, and IR spectra revealed the formation of a bilayer structure of 2 in which the mols. of trans-1 are dissolved to interact with the polar head groups. The photoreactivity of trans-1 in the cast film was found to be dramatically suppressed by heating at 40-60.degree.C, which is attributed to an improvement in the crystallinity of the bilayer.
- RE.CNT 49 THERE ARE 49 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

REFERENCE 6

- AN 128:101875 CA
- TI Photodimerization of cinnamic acids controlled by molecular assemblies of surfactant amine N-oxides
- AU Nakamura, Tomotaka; Takagi, Katsuhiko; Itoh, Mariko; Fujita, Kaori; Katsu, Hiroyuki; Imae, Toyoko; Sawaki, Yasuhiko
- CS Dep. Appl. Chem., Fac. Eng., Nagoya Univ., Nagoya, 464-01, Japan
- Journal of the Chemical Society, Perkin Transactions 2: Physical Organic Chemistry (1997), (12), 2751-2755 CODEN: JCPKBH; ISSN: 0300-9580
- PB Royal Society of Chemistry
- DT Journal
- LA English
- UV irradn. of (E)-4-RC6H4CH:CHCO2H [I, R = H, OMe] in a complex with AΒ surfactant amine N-oxides Me(CH2)nN(O)Me2 (II, n = 11, 13, 15) as vesicles in water leads to the formation of cyclodimers, i.e., .beta.-, .delta.-truxinic and/or .alpha.-truxillic acids. Decreasing the molar ratios of I to II causes the vesicles to transform into rod-like micelles and the yield of the cyclodimers decreases. The addn. of HCl or NaOH to aq. solns. of I and II brings about sharp changes in the self-assembly structures from vesicles to micro emulsions or rod-like micelles, accompanied by a change in product distribution in the photolysis of the complex. Upon addn. of photoinactive carboxylic acids, phenylpropionic and palmitic acids to the I-Me(CH2)16N(O)Me system, the rod-like micelles change into vesicles by formation of a complex with Me(CH2)16N(O)Me leading to the observation of a diln. effect in the photodimerization upon addn. of phenylpropionic acid. However, no diln. effect is obsd. for palmitic acid. This is found to be attributable to the difference in degrees of mixing of I with the acids in the vesicles. These results show

that photodimerization of I incorporated in II is controlled by a variety of mol. assemblies, i.e. rod-like micelles and homogeneous or heterogeneous vesicles.

RE.CNT 48 THERE ARE 48 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

REFERENCE 7

- AN 126:7719 CA
- TI Stereoselectivity control of [2 + 2] photocycloaddition by changing site distances of hydrotalcite interlayers
- AU Shichi, Tetsuya; Takagi, Katsuhiko; Sawaki, Yasuhiko
- CS Fac. Eng., Nagoya Univ., Nagoya, 464-01, Japan
- SO Chemical Communications (Cambridge) (1996), (17), 2027-2028 CODEN: CHCOFS; ISSN: 1359-7345
- PB Royal Society of Chemistry
- DT Journal
- LA English
- AB Stereoselectivity of photocyclodimers of unsatd. carboxylates is shown to be controlled by changing the site distances of clay interlayers.

REFERENCE 8

- AN 125:300268 CA
- TI Organized photocycloaddition of 4-benzoylbenzoate with unsaturated carboxylates in hydrotalcite clay interlayers
- AU Shichi, Tetsuya; Takagi, Katsuhiko; Sawaki, Yusuhiko
- CS Dep. of Applied Chem., Nagoya Univ., Nagoya, 464-01, Japan
- SO Chemistry Letters (1996), (9), 781-782 CODEN: CMLTAG; ISSN: 0366-7022
- PB Nippon Kagakkai
- DT Journal
- LA English
- AB Photoirradn. of a mixt. of 4-benzoylbenzoate (1) and 4-(2-phenylethenyl)benzoate yielded regioselective oxetanes in the presence of hydrotalcite clay. Similar irradn. of a mixt. of 1 and cinnamate gave rise to .beta.-truxinate, but not oxetane. The unexpected formation of oxetanes is understood by the favorable distance between C:O and C:C bonds packed in clay interlayers.

REFERENCE 9

- AN 125:33265 CA
- TI Photochemical cyclodimerization of cinnamic acids included in surfactant amine oxides
- AU Takagi, K.; Nakamura, T.; Katsu, H.; Itoh, M.; Sawaki, Y.; Imae, T.
- CS Dep. Appl. Chem., Fac. Eng., Nagoya Univ., Nagoya, 464-01, Japan
- SO Molecular Crystals and Liquid Crystals Science and Technology, Section A: Molecular Crystals and Liquid Crystals (1996), 277, 495-498
 CODEN: MCLCE9; ISSN: 1058-725X
- PB Gordon & Breach
- DT Journal
- LA English
- AB Photocyclodimerization of unsubstituted and p-methoxy substituted cinnamic acids incorporated in micelles, vesicles or microemulsions formed by dodecyl- and hexadecyldimethylamine oxides has been studied in water. The dimerization proceeds in vesicles much more efficiently than in micelles with preferable formation of head-to-head dimers. The photoreactivity and the stereochem. of the cyclodimers are affected by structural changes of the mol. aggregates.

REFERENCE 10

- AN 124:75179 CA
- TI Separation and identification of some monomeric and dimeric phenolic acids by a simple gas chromatographic method using a capillary column and FID-MSD

- AU Packert, M.; Steinhart, H.
- CS Inst. Biochemie Lebensmittelchemie, Univ. Hamburg, Hamburg, D-20146, Germany
- SO Journal of Chromatographic Science (1995), 33(11), 631-9 CODEN: JCHSBZ; ISSN: 0021-9665
- PB Preston Publications
- DT Journal
- LA English
- The monomeric and dimeric phenolic acids represent highly various AB structural substance classes. Included among the monomers are substituted benzoic acid, phenylacetic acid, phenylpropionic acid, and cinnamic acid; compds. of the biphenyl and cyclobutane types are representatives of dimeric acids. The formation of dimeric acids with a diphenylcyclobutane dicarboxylic acid structure from t-p-coumaric acid (t-p-CA) and t-ferulic acid (t-FA) via a [2+2]-photocycloaddn. reaction is thoroughly described. The products are characterized by using mass spectroscopy and by observing their chromatog. behavior. The applied method makes it possible to det. over 50 phenolic acid isomers. The following elution sequence for the monomeric acids is found:benzoic acids < phenylacetic acids < phenylpropionic acids < cinnamic acids. Among the dimeric acids, which appeared significantly after the monomeric acids, those with a diphenylcyclobutane dicarboxylic acid structure elute before cis/trans-dehydrodiferulic acid (c-t-DFA).

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L10 ANSWER 1 OF 36 CA COPYRIGHT 2003 ACS

AN 138:39088 CA

TI Topochemical reaction control in solution. III. 4,15-Diamino[2.2]paracyclophane, a reusable template for topochemical reaction control in solution

AU Zitt, Holger; Dix, Ina; Hopf, Henning; Jones, Peter G.

CS Institut fur Organische Chemie, Technische Universitat Braunschweig, Braunschweig, 38106, Germany

SO European Journal of Organic Chemistry (2002), (14), 2298-2307 CODEN: EJOCFK; ISSN: 1434-193X

PB Wiley-VCH Verlag GmbH

DT Journal

LA English

OS CASREACT 138:39088

An efficient synthesis of [2.2]paracyclophane-4,15-dicarboxylic acid from [2.2]paracyclophane (I) has been developed. The diacid was converted via a diazide into the 4,15-diisocyanato[2.2]paracyclophane (II), a versatile intermediate that could be transformed into many new pseudogeminally substituted derivs. of I. For example, treatment of II with alcs. provided two carbamates. On treatment of II with diisopropylamine, a urea compd., was obtained, whereas redn. with lithium aluminum hydride afforded a cyclic urea (III). Hydrolysis of II furnished a diamine (IV), which was used as a reusable spacer in a [2+2]photoaddn. expt. Thus, treatment of IV with trans-cinnamoyl chloride provided a bis(amide), which on irradn. in acetone ring-closed to give a cyclobutane compd. Sapon. of this yielded 3,4-diphenyl-1,2-cyclobutanedicarboxylic acid (.beta.-truxinic acid) and returned the spacer system 19, both in quant. yield. The X-ray structures of II and III are reported.

RE.CNT 37 THERE ARE 37 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

L10 ANSWER 2 OF 36 CA COPYRIGHT 2003 ACS

AN 135:117221 CA

TI Diphenylcyclobutane dicarboxylate derivatives as analgesics and antiinflammatory agents

IN Nakamura, Motoyuki; Yoshizawa, Toyokichi; Chitama, Akira; Nohara, Toshihiro; Sakurada, Shinobu

PA Showa Pharmaceutical Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 8 pp. CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

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GI

- AB Diphenylcyclobutane dicarboxylate derivs. (I; X1, X2, Y1, Y2, Z1, Z2 = H, OH, halogen, alkyl, alkoxy, N; R1, R2 = OH, halogen, alkoxyl, aryloxy, turpenoxy, sugar, N) are claimed as analgesics and antiinflammatory agents with min. toxicity. I were prepd., and their analgesic and antiinflammatory actions and effect on gastric mucosa were tested.
- L10 ANSWER 3 OF 36 CA COPYRIGHT 2003 ACS
- AN 133:321558 CA
- TI Compelled Orientational Control of the Solid-State Photodimerization of trans-Cinnamamides: Dicarboxylic Acid as a Non-covalent Linker
- AU Ito, Y.; Hosomi, H.; Ohba, S.
- CS Graduate School of Engineering, Department of Synthetic Chemistry and Biological Chemistry, Kyoto University, Kyoto, 606-8501, Japan
- SO Tetrahedron (2000), 56(36), 6833-6844 CODEN: TETRAB; ISSN: 0040-4020
- PB Elsevier Science Ltd.
- DT Journal
- LA English
- AΒ The 2:1 hydrogen-bonded cocrystals la.cntdot.ox, la.cntdot.su, la.cntdot.pht, la.cntdot.fu, lb.cntdot.ox, lc.cntdot.ox, ld.cntdot.ox between 4-R-trans-cinnamamides [1a-11c: R = H, Me, Cl, resp.; 1d = trans-3-(2-thienyl)acrylamide] and dicarboxylic acids (ox, su, gl, fu, pht) were prepd. and characterized by IR and powder X-ray techniques. crystal structures of la.cntdot.pht, la.cntdot.ox and la.cntdot.fu were solved by single crystal X-ray diffraction. Phthalic acid (pht) caused .beta.-type photodimerization of trans-cinnamamide (1a) in the cocrystal and functioned as a non-covalent linker like gauche 1,2-diamines in photodimerization of trans-cinnamic acids. Oxalic acid (ox) forced la to take a bilayer structure that is suitable for .beta.-type photodimerization. In the case of fumaric acid (fu), cross photodimerization with la occurred to give a cycloadduct. For the cocrystals la.cntdot.pht and la.cntdot.fu, pedal-like motion was assumed to occur prior to the dimerization.
- RE.CNT 37 THERE ARE 37 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT
- L10 ANSWER 4 OF 36 CA COPYRIGHT 2003 ACS
- AN 129:148662 CA
- TI A controlled photodimerization of cinnamic acid in a composite bilayer of dioctadecyldimethylammonium bromide
- AU Nakamura, Tomotaka; Takagi, Katsuhiko; Sawaki, Yasuhiko
- CS Department of Applied Chemistry, Graduate School of Engineering, Nagoya University, Nagoya, 464-01, Japan
- SO Molecular Crystals and Liquid Crystals Science and Technology, Section A: Molecular Crystals and Liquid Crystals (1998), 313, 341-346 CODEN: MCLCE9; ISSN: 1058-725X
- PB Gordon & Breach Science Publishers
- DT Journal
- LA English
- AB Investigations of the photochem. of cinnamic acid (CA) were carried out in a form of 1: 1 composite solids with dioctadecyldimethylammonium bromide (DODAB) or chloride (DODAC). UV irradn. of the composite solids with DODAB resulted in the highly selective formation of the syn-Head-to-Head dimer, suggesting the parallel orientation of CA within the complex. Studies of X-ray diffraction, DSC, and IR spectra revealed the formation

of a composite bilayer assembly showing the interaction between polar groups of CA and DODAB as well as the incorporation of CA into the alkyl regions of DODAB. On the other hand, the mixt. of CA and DODAC showed less and non-selective photodimerization reaction, which could be ascribed to a disordered mol. arrangement, judging from a weak and broad endothermic peak at around reaction temp. in the DSC curve.

RE.CNT 24 THERE ARE 24 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

- L10 ANSWER 5 OF 36 CA COPYRIGHT 2003 ACS
- AN 129:15910 CA
- TI Photodimerization of trans-cinnamic acid in a bilayer of dimethyldioctadecylammonium bromide
- AU Nakamura, Tomotaka; Takagi, Katsuhiko; Sawaki, Yasuhiko
- CS Dep. Applied Chemistry, Fac. Eng., Nagoya Univ., Nagoya, 464-01, Japan
- SO Bulletin of the Chemical Society of Japan (1998), 71(4), 909-914 CODEN: BCSJA8; ISSN: 0009-2673
- PB Chemical Society of Japan
- DT Journal
- LA English
- AB The photochem. of trans-cinnamic acid (trans-1) was studied in a cast film of dimethyldioctadecylammonium bromide (2), deposited from aq. dispersions as a cryst. solid by the spontaneous vaporization of water. UV irradn. of the cast film resulted in a highly selective formation of the syn-Head-to-Head dimer, suggesting a parallel orientation of trans-1 within a bilayer film. Analyses with X-ray diffraction, differential scanning calorimetry, and IR spectra revealed the formation of a bilayer structure of 2 in which the mols. of trans-1 are dissolved to interact with the polar head groups. The photoreactivity of trans-1 in the cast film was found to be dramatically suppressed by heating at 40-60.degree.C, which is attributed to an improvement in the crystallinity of the bilayer.
- RE.CNT 49 THERE ARE 49 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT
- L10 ANSWER 6 OF 36 CA COPYRIGHT 2003 ACS
- AN 128:101875 CA
- TI Photodimerization of cinnamic acids controlled by molecular assemblies of surfactant amine N-oxides
- AU Nakamura, Tomotaka; Takagi, Katsuhiko; Itoh, Mariko; Fujita, Kaori; Katsu, Hiroyuki; Imae, Toyoko; Sawaki, Yasuhiko
- CS Dep. Appl. Chem., Fac. Eng., Nagoya Univ., Nagoya, 464-01, Japan
- SO Journal of the Chemical Society, Perkin Transactions 2: Physical Organic Chemistry (1997), (12), 2751-2755
 CODEN: JCPKBH; ISSN: 0300-9580
- PB Royal Society of Chemistry
- DT Journal
- LA English
- AΒ UV irradn. of (E)-4-RC6H4CH:CHCO2H [I, R = H, OMe] in a complex with surfactant amine N-oxides Me(CH2)nN(O)Me2 (II, n = 11, 13, 15) as vesicles in water leads to the formation of cyclodimers, i.e., .beta.-, .delta.-truxinic and/or .alpha.-truxillic acids. Decreasing the molar ratios of I to II causes the vesicles to transform into rod-like micelles and the yield of the cyclodimers decreases. The addn. of HCl or NaOH to aq. solns. of I and II brings about sharp changes in the self-assembly structures from vesicles to micro emulsions or rod-like micelles, accompanied by a change in product distribution in the photolysis of the complex. Upon addn. of photoinactive carboxylic acids, phenylpropionic and palmitic acids to the I-Me(CH2)16N(O)Me system, the rod-like micelles change into vesicles by formation of a complex with Me(CH2)16N(O)Me leading to the observation of a diln. effect in the photodimerization upon addn. of phenylpropionic acid. However, no diln. effect is obsd. for palmitic acid. This is found to be attributable to the difference in degrees of mixing of I with the acids in the vesicles. These results show that photodimerization of I incorporated in II is controlled by a variety of mol. assemblies, i.e. rod-like micelles and homogeneous or heterogeneous vesicles.
- RE.CNT 48 THERE ARE 48 CITED REFERENCES AVAILABLE FOR THIS RECORD

ALL CITATIONS AVAILABLE IN THE RE FORMAT

- L10 ANSWER 7 OF 36 CA COPYRIGHT 2003 ACS
- 126:7719 CA ΑN
- Stereoselectivity control of [2 + 2] photocycloaddition by changing site TI distances of hydrotalcite interlayers
- Shichi, Tetsuya; Takagi, Katsuhiko; Sawaki, Yasuhiko ΑU
- CS
- Fac. Eng., Nagoya Univ., Nagoya, 464-01, Japan Chemical Communications (Cambridge) (1996), (17), 2027-2028 SO CODEN: CHCOFS; ISSN: 1359-7345
- PB Royal Society of Chemistry
- DTJournal
- English LA
- AB Stereoselectivity of photocyclodimers of unsatd. carboxylates is shown to be controlled by changing the site distances of clay interlayers.
- L10 ANSWER 8 OF 36 CA COPYRIGHT 2003 ACS
- AN 125:300268 CA
- TIOrganized photocycloaddition of 4-benzoylbenzoate with unsaturated carboxylates in hydrotalcite clay interlayers
- Shichi, Tetsuya; Takagi, Katsuhiko; Sawaki, Yusuhiko ΑIJ
- CS Dep. of Applied Chem., Nagoya Univ., Nagoya, 464-01, Japan
- SO Chemistry Letters (1996), (9), 781-782 CODEN: CMLTAG; ISSN: 0366-7022
- PR Nippon Kagakkai
- DT Journal
- LAEnglish
- AΒ Photoirradn. of a mixt. of 4-benzoylbenzoate (1) and 4-(2phenylethenyl)benzoate yielded regioselective oxetanes in the presence of hydrotalcite clay. Similar irradn. of a mixt. of 1 and cinnamate gave rise to .beta.-truxinate, but not oxetane. The unexpected formation of oxetanes is understood by the favorable distance between C:O and C:C bonds packed in clay interlayers.
- L10ANSWER 9 OF 36 CA COPYRIGHT 2003 ACS
- AN125:33265 CA
- TIPhotochemical cyclodimerization of cinnamic acids included in surfactant amine oxides
- ΑIJ Takagi, K.; Nakamura, T.; Katsu, H.; Itoh, M.; Sawaki, Y.; Imae, T.
- CS Dep. Appl. Chem., Fac. Eng., Nagoya Univ., Nagoya, 464-01, Japan
- Molecular Crystals and Liquid Crystals Science and Technology, Section A: SO Molecular Crystals and Liquid Crystals (1996), 277, 495-498 CODEN: MCLCE9; ISSN: 1058-725X
- PB Gordon & Breach
- DTJournal
- LΑ English
- AB Photocyclodimerization of unsubstituted and p-methoxy substituted cinnamic acids incorporated in micelles, vesicles or microemulsions formed by dodecyl- and hexadecyldimethylamine oxides has been studied in water. dimerization proceeds in vesicles much more efficiently than in micelles with preferable formation of head-to-head dimers. The photoreactivity and the stereochem. of the cyclodimers are affected by structural changes of the mol. aggregates.
- L10 ANSWER 10 OF 36 CA COPYRIGHT 2003 ACS
- AN 124:75179 CA
- Separation and identification of some monomeric and dimeric phenolic acids TΤ by a simple gas chromatographic method using a capillary column and FID-MSD
- ΑU Packert, M.; Steinhart, H.
- CS Inst. Biochemie Lebensmittelchemie, Univ. Hamburg, Hamburg, D-20146,
- SO Journal of Chromatographic Science (1995), 33(11), 631-9 CODEN: JCHSBZ; ISSN: 0021-9665
- PΒ Preston Publications
- DTJournal
- LA English

- AΒ The monomeric and dimeric phenolic acids represent highly various structural substance classes. Included among the monomers are substituted benzoic acid, phenylacetic acid, phenylpropionic acid, and cinnamic acid; compds. of the biphenyl and cyclobutane types are representatives of dimeric acids. The formation of dimeric acids with a diphenylcyclobutane dicarboxylic acid structure from t-p-coumaric acid (t-p-CA) and t-ferulic acid (t-FA) via a [2+2]-photocycloaddn. reaction is thoroughly described. The products are characterized by using mass spectroscopy and by observing their chromatog. behavior. The applied method makes it possible to det. over 50 phenolic acid isomers. The following elution sequence for the monomeric acids is found:benzoic acids < phenylacetic acids < phenylpropionic acids < cinnamic acids. Among the dimeric acids, which appeared significantly after the monomeric acids, those with a diphenylcyclobutane dicarboxylic acid structure elute before cis/trans-dehydrodiferulic acid (c-t-DFA).
- L10 ANSWER 11 OF 36 CA COPYRIGHT 2003 ACS
- AN 123:338809 CA
- TI Control of solid-state photodimerization of trans-cinnamic acid by double salt formation with diamines
- AU Ito, Yoshikatsu; Borecka, Bozena; Trotter, Majes; Scheffer, John R.
- CS Fac. Eng., Kyoto Univ., Kyoto, 606, Japan
- SO Tetrahedron Letters (1995), 36(34), 6083-6 CODEN: TELEAY; ISSN: 0040-4039
- PB Elsevier
- DT Journal
- LA English
- AB By double salt formation, diamines can steer the solid-state [2+2] photodimerization of trans-cinnamic acid (1). Thus, the yields for the photodimerization were significant only in three double salts, i.e., the ones with 1,3-propanediamine and trans- and cis-1,2-cyclohexanediamine, which are assumed to have an overlap structure. The result photodimer is generally .beta.-truxinic acid although in one case, .epsilon.-truxillic acid was formed. .alpha.-Truxillic acid was not produced from any of the double salts studied. The X-ray crystal structures of three double salts of low photoreactivity (1.cntdot.ethylenediamine, 1.cntdot.1,2-propanediamine, 1.cntdot.1,6-hexanediamine) were consistent with Schmidt's rule.
- L10 ANSWER 12 OF 36 CA COPYRIGHT 2003 ACS
- AN 122:326222 CA
- TI Photocyclodimerization of Cinnamic Acid on a Reaction Matrix: Structural Effect of Molecular Assemblies Constructed by Amphiphilic Compounds
- AU Imae, Toyoko; Tsubota, Toshinari; Mori, Osamu; Takagi, Katsuhiko; Itoh, Mariko; Sawaki, Yasuhiko
- CS Department of Chemistry, Faculty of Science, Chikusa, 464, Japan
- SO Journal of Physical Chemistry (1995), 99(16), 6046-53 CODEN: JPCHAX; ISSN: 0022-3654
- PB American Chemical Society
- DT Journal
- LA English
- AΒ Light scattering measurement, video-enhanced differential interference microscopic and electron microscopic observations, 1H-NMR measurement, and photocyclodimerization were carried out for aq. solns. of alkyldimethylamine oxides (C12DAO and C16DAO) and dioctadecyldimethylammonium chloride (2C18DAC) with addn. of cinnamic acid. Mol. assemblies in CnDAO: cinnamic acid solns. were converted from micelles to vesicles with increasing mixing ratios of cinnamic acid, while 2C18DAC in water always formed vesicles with addn. of cinnamic acid. Structural changes in reaction matrixes and mol. assemblies constructed by amphiphilic compds. were reflected in the reactivity and stereochem. selectivity of photocyclodimerization. It was concluded that the photodimerization proceeded effectively in vesicles rather than in micelles. The dimerization yields increased with mixing ratio and were in the order of syn-HT < syn-HH < anti-HH (where HT represents head-to-tail and HH represents head-to-head). The product selectivity was discussed in relation to the arrangement of cinnamic acid on mol. assemblies.

L10 ANSWER 13 OF 36 CA COPYRIGHT 2003 ACS

AN 122:150816 CA

TI .epsilon.-PMTC, an opioid antagonist bivalent ligand based on N-(phenylpropyl)normetazocine with very high preference for .mu. sites

AU Ronsisvalle, Giuseppe; Pappalardo, Maria S.; Prezzavento, Orazio; Marrazzo, Agostino; Caccamese, Salvatore; Carboni, Lucia; Spampinato,

CS Istituto di Chimica Farmaceutica e Tossicologica, Universita degli Studi di Catania, Catania, 95125, Italy

SO Medicinal Chemistry Research (1994), 4(4), 245-53 CODEN: MCREEB; ISSN: 1054-2523

PB Birkhaeuser

DT Journal

LA English

GI

HO Me
$$Ph$$
 CH_2 Ph H OH Me Me Me Me

AB The synthesis and opioid receptor affinity of three rigid isomeric bivalent ligands consisting of a cyclobutane matrix that supports two pharmacophoric units derived from benzomorphan are described. The bivalent ligand cis-1,3-bis[3-(6,11-dimethyl-1,2,3,4,5,6-hexahydro-2,6-methano-8-hydroxy-3-benzazocinyl)methyl]- cis-2,4-diphenyl-cyclobutane, (I .epsilon.-PMTC) displayed selective binding affinity toward .mu. opioid receptors and antagonist character in both in vivo and in vitro assays.

Ι

L10 ANSWER 14 OF 36 CA COPYRIGHT 2003 ACS

AN 121:267525 CA

TI Aggregate Formation and Photoreactivity in Phospholipid Vesicles and Langmuir-Blodgett Multilayers: Topologically Controlled Photodimerization of Amphiphilic Styrenes

AU Zhao, Xiao-Mei; Perlstein, Jerome; Whitten, David G.

CS Department of Chemistry, University of Rochester, Rochester, NY, 14627, USA

SO Journal of the American Chemical Society (1994), 116(23), 10463-7 CODEN: JACSAT; ISSN: 0002-7863

DT Journal

LA English

AB The amphiphilic styrene fatty acid (I) and its corresponding phosphatidylcholine deriv. (II) have been found to exhibit aggregate formation and topol. controlled photodimerization in Langmuir-Blodgett (LB) assemblies and phospholipid vesicles, resp. For both compds., formation of the .beta. or syn, head-to-head photodimer is the chief photoreaction obsd. upon irradn. in the assemblies; a trace of the cis styrene is obsd. on irradn. in the phospholipid vesicles. Changes in the absorption spectrum and a lack of fluorescence in the assemblies, compared to dil. org. solns., suggest that I and II exist as "H" aggregates in both the LB films and vesicles. Monte Carlo simulations suggest that the most stable structure in a monolayer is a simple translation aggregate. Energy minimization gives a cross sectional area in good agreement with that measured for films at the air-water interface. The calcd. structure has nearest neighbor sepn. within the "magic distance" for solid state

photodimerization of 4.10 .ANG.. The relatively small calcd. tilt angle suggests that dimers can be formed without substantial reorganization. The .lambda.max for the aggregate estd. by an extended dipole calcn. shows good agreement with the measured absorption max. Vesicles of II, either pure or with excess satd. phospholipid, give topol. controlled formation of the .beta. dimer via photolysis of an aggregated form.

- L10 ANSWER 15 OF 36 CA COPYRIGHT 2003 ACS
- AN 120:263170 CA
- TI The anatomy of a cocaine comparison case: a prosecutorial and chemistry perspective
- AU Moore, James M.; Meyers, Richard P.; Jimenez, Marcos D.
- CS Spec. Test. Res. Lab., U.S. Drug Enforcement Adm., McLean, VA, USA
- SO Journal of Forensic Sciences (1993), 38(6), 1305-25 CODEN: JFSCAS; ISSN: 0022-1198
- DT Journal
- LA English
- AΒ Methodol. used for the comparative chem. analyses of two illicit cocaine seizures, and its application in a successful criminal prosecution, is described. A description of events leading to the arrest of the defendant and an overview of the jury trial are provided. Illicit cocaine, found in the defendant's suitcase and wallet, was subjected to chem. derivatization and three distinct gas chromatog. methods for the detection and relative quantitation of cocaine manufg. impurities/byproducts. The cocaine impurities included cis- and trans-cinnamoylcocaine, the isomeric truxillines and the hydroxycocaines. Among the cocaine manufg. byproducts detected were benzoylecgonine, ecgonine Me ester, ecgonine, N-benzoylnorecgonine Me ester and N-norcocaine. Chem. derivatization of the cocaine samples was accomplished using heptafluorobutyric anhydride and N,O-bis(trimethylsilyl)acetamide. The derivatized impurities/byproducts were subjected to capillary gas chromatog. anal. using both flame ionization and electron-capture detectors. The comparative chem. analyses provided a pos. correlation between the suitcase and wallet cocaine samples.
- L10 ANSWER 16 OF 36 CA COPYRIGHT 2003 ACS
- AN 118:124269 CA
- TI Synthesis of a 7,7'-cyclolignane and a 7',8-cyclo-7,8'-neolignane via photodimerization of (E)-cinnamic acids in the solid state
- AU Fonseca, Sebastiao F.; Rossi, Eleonilce Rosa B.
- CS Inst. Quim., Univ. Estadual Campinas, Campinas, 13081, Brazil
- SO Journal of Chemical Research, Synopses (1992), (12), 406-7 CODEN: JRPSDC; ISSN: 0308-2342
- DT Journal
- LA English

GΙ

AB The photodimerizations of (E)-3,4-RR1C6H3CH:CHCO2H (R,R1 = methylenedioxy, R = R1 = MeO) in the solid state, followed by a short sequence of transformations, afforded 7,7'-cyclolignane I and 7',8-cyclo-7,8'-neolignane II resp.

L10 ANSWER 17 OF 36 CA COPYRIGHT 2003 ACS

AN 116:214025 CA

TI Stereocontrolled photochemical [2 + 2] cycloaddition

AU Fleming, Steven A.; Ward, Susan C.

CS Dep. Chem., Brigham Young Univ., Provo, UT, 84602, USA

SO Tetrahedron Letters (1992), 33(8), 1013-16

CODEN: TELEAY; ISSN: 0040-4039

DT Journal

LA English

OS CASREACT 116:214025

AB Siloxanes were used as tethers in photochem. [2+2] reactions.

Dicinnamyldimethylsiloxane and dicinnamyldiphenylsiloxane were prepd. from cinnamyl alc. and the resp. silanes. After photolysis only one cyclobutane product, the all trans isomer, was obsd. in nearly quant. yield. Titanium as tether gave very low conversions.

L10 ANSWER 18 OF 36 CA COPYRIGHT 2003 ACS

AN 111:194131 CA

TI New host family based on small-ring compounds

AU Weber, Edwin; Hecker, Manfred; Csoeregh, Ingeborg; Czugler, Matyas

CS Inst. Org. Chem. Biochem., Univ. Bonn, Bonn, D-5300/1, Fed. Rep. Ger.

SO Journal of the American Chemical Society (1989), 111(20), 7866-72

CODEN: JACSAT; ISSN: 0002-7863

DT Journal

LA English

OS CASREACT 111:194131

GI

AB Three- and four-membered ring compds. with functional groups and bulky substituents, e.g., I, II and III, have proved to be a rewarding new source of inclusion hosts. These hosts form inclusion complexes with a variety of uncharged org. mols. ranging from protic dipolar to apolar compds. (168 different inclusion species). Formation and selectivity depend in a systematic manner on structural parameters of the host, such as the nature, no., and position of functional groups, the substituents, and ring size. X-ray structure analyses of I, I.Me3COH and II.MeCN are reported, revealing the building principles of the new clathrate family. The structures exhibit linear chains of inter-/intramol. H-bridges between carboxylic groups in the free host I and H-bridge aggregation of host and guest mols. in infinite helical chains for the I.Me3COH (1:1) inclusion. In II.MeCN (1:1), the guest mols. are tightly enclosed by the host framework without further specific interactions.

L10 ANSWER 19 OF 36 CA COPYRIGHT 2003 ACS

AN 110:23458 CA.

TI Regioselective photodimerization of cinnamic acid in a micellar solution

AU Nakamura, Yushin

CS. Fac. Pharm. Sci., Josai Univ., Sakado, 350-02, Japan

Journal of the Chemical Society, Chemical Communications (1988), (7),
477-8
CODEN: JCCCAT; ISSN: 0022-4936

DT Journal

LA English

AB Irradn. of trans-cinnamic acid (I) and its Me ester (II) in 1% aq. CTAB gave dimeric products in 35 and 29% yields as a mixt. of truxinic and truxillic acids in a ratio of 19:1 and 3.8:1, resp., while in homogeneous solns. I gave no photodimers and II gave dimeric products in 2.3% yield under similar reaction conditions.

L10 ANSWER 20 OF 36 CA COPYRIGHT 2003 ACS

AN 107:192404 CA

TI Analysis of manufacturing by-products and impurities in illicit cocaine via high-performance liquid chromatography and photodiode array detection

AU Lurie, I. S.; Moore, J. M.; Cooper, D. A.; Kram, T. C.

CS Spec. Test. Res. Lab., Drug Enforc. Adm., McLean, VA, 22102, USA

SO Journal of Chromatography (1987), 405, 273-81

CODEN: JOCRAM; ISSN: 0021-9673

DT Journal

LA English

GI

Ι

AΒ An HPLC method is reported for the detection of manufq. byproducts and impurities in illicitly produced cocaine (I). For the 1st time, peak-enriched chromatograms were obtained using HPLC, and were accomplished using reversed-phase chromatog, and photodiode array detection. The use of SDS as an ion-pairing reagent permitted the simultaneous sepn. of acids, monoprotic amines, and diprotic amines, and, in combination with gradient elution, greatly increased the no. of compds. sepd. A mixed binary-ternary gradient was used to further optimize the Dual UV detection at 215 and 277 nm was used. The chromatogram at 215 nm consisted 1st of carboxylic acids such as benzoic acid, cinnamic acid (cis and trans), and several isomers of truxillic and truxinic acid; next, the monoprotic amines benzoylecgonine and cinnamoylcocaine (cis and trans); and last a group of compds. which are believed to be isomers of the diprotic amine truxilline. In addn., simultaneous detection at 277 nm permitted the selective detection of various compds., some of which are addnl. components. The rapid acquisition of UV spectra greatly facilitated compd. identification.

L10 ANSWER 21 OF 36 CA COPYRIGHT 2003 ACS

AN 107:153689 CA

TI Organic photochemical reactions. 81. Conrotatory electrocycloreversion of cis-3,4-diarylcyclobutene cation radical

AU Miyashi, Tsutomu; Wakamatsu, Kan; Akiya, Teruhiko; Kikuchi, Koichi; Mukai, Toshio

CS Fac. Sci., Tohoku Univ., Sendai, 980, Japan

SO Journal of the American Chemical Society (1987), 109(17), 5270-1 CODEN: JACSAT; ISSN: 0002-7863

DT Journal

LA English

OS CASREACT 107:153689

AB In order to test theor. predications for a stereochem. course of the electrocycloreversion reaction of a cyclobutene radical cation to a 1,3-butadiene radical cation, photochem. electron-transfer reactions of cis-3,4-diarylcyclobutenes were investigated. Cis-3,4-Diarylcyclobutene radical cations generated either by the photoexcitation of the electron donor-acceptor complexes with TCNE or under the 2,6,9,10-tetracyanoanthracene-sensitized conditions efficiently isomerized to cis,trans-1,4-diaryl-1,3-butadienes, demonstrating the occurrence of the stereospecific conrotatory electrocycloreversion.

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L10 ANSWER 22 OF 36 CA COPYRIGHT 2003 ACS
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AN 106:66920 CA

TI Cyclobutane compounds

CASREACT 106:66920

IN Nishikubo, Tatatomi

PA Research Development Corp. of Japan, Japan

SO Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF

DT Patent

LA /Japanese

FAN CNT 1

OS

GI

TILTIN 9	CNII					
1/	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE	
,						
ΡI	JP 61212532	A2	19860920	JP 1985-53392	19850319	
	JP 05088210	B4	19931221			
PRAI	JP 1985-53392		19850319			

CO2R I

AB Cyclobutane derivs. (I; R = p-O2NC6H4, 4-nitro-1-naphthyl, etc.), useful as optical materials, are prepd. by photocycloaddn. of PhCH:CHCO2R (II) in the presence of a dispersant solvent. Thus, 0.059 mol II (R = p-O2NC6H4) in a mixed solvent of 380 mL and 20 mL hexane-MeOH at a dispersion of 20 g/L was irradiated 6 h at 30.degree. with stirring to give 91.2% I (R = p-O2NC6H4).

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L10 ANSWER 23 OF 36 CA COPYRIGHT 2003 ACS
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AN 105:152559 CA

TI Convenient synthesis of .beta.-truxinic acid via photodimerization of p-nitrophenyl cinnamate in the crystalline state

AU Nishikubo, Tadatomi; Takahashi, Eiji; Miyaji, Takatoshi; Iizawa, Takashi

CS Fac. Eng., Kanagawa Univ., Yokohama, 221, Japan

SO Bulletin of the Chemical Society of Japan (1985), 58(11), 3399-400 CODEN: BCSJA8; ISSN: 0009-2673

DT Journal

LA English

OS CASREACT 105:152559

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- AB .beta.-Truxinic acid (I, R = H) was prepd. in high yield via topochem. photodimerization of PhCH:CHCO2C6H4NO2-4 (II) crystals, followed by hydrolysis of the obtained dimer. Thus, II crystals were suspended in hexane under N and irradiated with a 100-W high pressure Hg lamp to give 99.2% I (R = 4-O2NC6H4) which was treated with KOH in MeOH and then concd. HCl to give 96.7% I (R = H).
- L10 ANSWER 24 OF 36 CA COPYRIGHT 2003 ACS
- AN 103:203595 CA
- TI Application of anisotropic photocleavage of head-to-head type cyclobutane compounds
- AU Yonezawa, Noriyuki; Yamashita, Takashi; Kanoe, Toshio; Saigo, Kazuhiko; Hasegawa, Masaki
- CS Fac. Eng., Univ. Tokyo, Tokyo, 113, Japan
- SO Industrial & Engineering Chemistry Product Research and Development (1985), 24(4), 593-8 CODEN: IEPRA6; ISSN: 0196-4321
- DT Journal
- LA English
- AB The cyclobutane ring in head-to-head coumarin dimers and their lactone-opened derivs. photocleaves in 3 different modes: sym., asym., and competitive. On the basis of the structural factors governing these pathways, controlling the mode of the photocleavage reaction of some head-to-head cyclobutane compds. was put into practice. The polyamide derived from coumarin dimer was transformed to another linear polyamide by exclusively asym. photocleavage of the cyclobutane rings in the main chain. The pH of a soln. contg. monoanilide derivs. of coumarin dimer was changed by photoirradn. due to the cleavage of the cyclobutane ring followed by spontaneous relactonization. An unsym. substituted stilbene was synthesized from a substituted cinnamic anhydride by intramol. photocycloaddn. and asym. photocleavage.
- L10 ANSWER 25 OF 36 CA COPYRIGHT 2003 ACS
- AN 102:45264 CA
- TI Luminescence and photolytic cycloreversion of cyclobutane derivatives: cinnamic acid dimers and their diamides
- AU Aloisi, Gian Gaetano; Mazzucato, Ugo; Bartocci, Giampiero; Cavicchio, Giancarlo; Maravigna, Pietro; Montaudo, Giorgio
- CS Dip. Chim., Univ. Perugia, Perugia, I-06100, Italy
- SO Zeitschrift fuer Physikalische Chemie (Muenchen, Germany) (1983), 138(2), 207-21
 - CODEN: ZPCFAX; ISSN: 0044-3336
- DT Journal
- LA English
- AB The fluorescence and photocycloreversion quantum yields for four cinnamic acid dimers and their diamides have been detd. in acetonitrile and methanol and their decay rate parameters estd. The results emphasize that the deactivation of the excited dimers occurs mainly through radiationless paths different from the photochem. cleavage reaction. The corresponding diamides have been also studied in water-methanol solns. at different pH. The trend of the fluorescence and photoreaction quantum yield as a function of pH gives information on the role of the acid-base equil. in the excited state on the photocleavage process.
- L10 ANSWER 26 OF 36 CA COPYRIGHT 2003 ACS
- AN 93:150844 CA
- TI Photodegradation of polyamides and model diamides containing photoreactive truxillic units
- AU Aloisi, Gian Gaetano; Mazzucato, Ugo; Maravigna, Pietro; Montaudo,

Giorgio; Recca, Antonio; Scamporrino, Emilio

- CS Dip. Chim., Univ. Perugia, Perugia, Italy
- SO Chimica e l'Industria (Milan, Italy) (1979), 61(11), 800-5 CODEN: CINMAB; ISSN: 0009-4315
- DT Journal
- LA English
- AB The photodegrdn. of polyamides and diamides of truxillic acid (I) and truxinic acid was studied by viscometry and spectrophotometry. In copolyamides of I and adipic acid, a relatively low I content gives a marked increase in the photolysis rate. The presence of strong acids increases the quantum yield from I derivs. (.alpha. and .epsilon. isomers) noticeably. The similarity of plots of fluorescence and photolysis vs. pH indicates that protonation in the 1st excited singlet state favors ring cleavage of these isomers. Photosensitization to shift the excitation spectrum of photolysis towards the visible region was studied.
- L10 ANSWER 27 OF 36 CA COPYRIGHT 2003 ACS
- AN 92:216000 CA
- TI Stereoselective syntheses via a photochemical template effect
- AU Damen, Julien; Neckers, D. C.
- CS Dep. Chem., Bowling Green State Univ., Bowling Green, OH, 43403, USA
- SO Journal of the American Chemical Society (1980), 102(9), 3265-7 CODEN: JACSAT; ISSN: 0002-7863
- DT Journal
- LA English
- AB Bis(vinylbenzyl) truxillate-divinylbenzene-styrene copolymers (I) were hydrolyzed, the polymer alcs. were esterified with trans-cinnamoyl chloride and UV-irradiated in C6H6 suspension, and the resulting photodimers were liberated by acid hydrolysis. The polymer matrix directed the dimerization to occur >50% of the time in a stereochem. direction which did not occur at all in monomers or in random polymer analogs. Irradn. of the trans-cinnamate of hydrolyzed I prepd. with .beta.-truxinic acid (II) [528-34-7] vinylbenzyl ester, for example, gave 53% II, although the cinnamate of hydrolyzed random I gave only the .alpha.-isomer.
- L10 ANSWER 28 OF 36 CA COPYRIGHT 2003 ACS
- AN 88:112479 CA
- TI Cathodic reduction of derivatives of dibenzylidenesuccinic acid; attempted electrohydrocyclization of conjugated systems
- AU Andersson, Jan; Eberson, Lennart
- CS Chem. Cent., Univ. Lund, Lund, Swed.
- SO Nouveau Journal de Chimie (1977), 1(5); 413-18 CODEN: NJCHD4; ISSN: 0398-9836
- DT Journal
- LA English
- AB Bisactivated conjugated diolefins, in the form of the (E,E) and (Z,Z) forms of dibenzylidenesuccinic acid and the (E,E) isomer of its N-methylimide, were reduced at a Hg cathode. The expected cyclobutane derivs. (di-Me truxinates and its N-methylimide) were however not formed; instead products of double bond satn., (derivs. of benzylbenzylidene as well as 2,3-dibenzylsuccinic acid) were isolated.
- L10 ANSWER 29 OF 36 CA COPYRIGHT 2003 ACS
- AN 86:16321 CA
- TI Solid state photodimerization of surfactant esters of cinnamic acid
- AU Bolt, John; Quina, Frank H.; Whitten, David G.
- CS Dep. Chem., Univ. North Carolina, Chapel Hill, NC, USA
- 30 Tetrahedron Letters (1976), (30), 2595-8
 - CODEN: TELEAY; ISSN: 0040-4039
- ্র Journal
- LA English

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QD 241 TYZ

- AB Octadecyl trans-cinnamate (I) on crystn. from several solvents under various conditions gave 3 different crystal forms. Form A, prepd. by slow crystn. from alcs. or by concg. solns. of I in hydrocarbon or chlorocarbon solvents, on irradn. dimerized to the diester II (R = H). Form B, obtained by crystn. of I from ketonic solvents, was photostable. Form C, obtained by rapid cooling of EtOH solns. of I, on irradn. dimerized to the diester III (R = H). Similarly, octadecyl trans-p-chlorocinnamate (IV) provided 2 crystal forms. Recrystn. of IV from Me2CO gave photostable crystals. Crystals obtained from 5:1 EtOH-C6H6 dimerized on irradn. to the diester II (R = Cl). Irradn. of IV at .apprx.40.degree. gave a mixt. of II and III (R = Cl).
- L10 ANSWER 30 OF 36 CA COPYRIGHT 2003 ACS
- AN 82:155084 CA
- TI Photodimers of cinnamic acid and related compounds. Stereochemical study of NMR
- AU Montaudo, G.; Caccamese, S.; Librando, V.
- CS Inst. Ind. Chem., Univ. Catania, Catania, Italy
- SO Organic Magnetic Resonance (1974), 6(10), 534-6 CODEN: ORMRBD; ISSN: 0030-4921
- DT Journal
- LA English
- AB The NMR spectra of photodimers of cinnamic acid, chalcone, and 9 related compds. were analyzed and all coupling consts. obtained. K,L,M, and N parameters allowed head-to-head (hh) and head-to-tail (ht) stereoisomers to be distinguished. The different types of hh and ht isomers were assigned on the basis of 3-bond couplings. An example of an A2B2 spin system was reported.
- L10 ANSWER 31 OF 36 CA COPYRIGHT 2003 ACS
- AN 82:154628 CA
- TI Photodimers of cinnamic acid and related compounds. Stereochemical study by electron-impact and field desorption mass spectrometry
- AU Caccamese, S.; Montaudo, G.; Przybylski, M.
- CS Inst. Ind. Chem., Univ. Catania, Catania, Italy
- SO Organic Mass Spectrometry (1974), 9(11), 1114-23 CODEN: ORMSBG; ISSN: 0030-493X
- DT Journal
- LA English
- AB The 13 eV electron-impact and field desorption mass spectra of photodimers of cinnamic acid and related compds. contg. a cyclobutane ring revealed both sym. and asym. splittings of the ring, thus allowing characterization of the head-to-head and head-to-tail structures of the dimers. A McLafferty type ring opening rearrangement was also obsd.
- L10 ANSWER 32 OF 36 CA COPYRIGHT 2003 ACS
- AN 82:9946 CA
- TI Photochemical scission of cinnamic acid dimers. .alpha.-Truxillic and .beta.-truxinic acids

- AU Rennert, J.; Grossman, D.
- CS City Coll., City Univ. New York, New York, NY, USA
- SO Journal of Photochemistry (1974), 3(2-3), 163-70 CODEN: JPCMAE; ISSN: 0047-2670
- DT Journal
- LA English
- AB Excited .alpha.-truxillic acid and .beta.-truxinic acid monomerize in soln. to yield cinnamic acid with quantum efficiencies of 0.12 and 0.41, resp. The photoscission proceeds via excited singlet and triplet mols.
- L10 ANSWER 33 OF 36 CA COPYRIGHT 2003 ACS
- AN 78:97032 CA
- TI Structure and conformation of chalcone photodimers and related compounds
- AU Montaudo, G.; Caccamese, S.
- CS Inst. Ind. Chem., Univ. Catania, Catania, Italy
- SO Journal of Organic Chemistry (1973), 38(4), 710-16 CODEN: JOCEAH; ISSN: 0022-3263
- DT Journal
- LA English
- GI For diagram(s), see printed CA Issue.
- Configurational assignments and the conformational preferences of the 2 AΒ chalcone (benzalacetophenone) photodimers were reported. Mass spectra and NMR data provided the evidence for the configurational assignments, while dipole moment data and conformational energy ests. were used in the conformational work. Detailed anal. of mass and NMR spectra allowed assignment of the .beta.-truxinic structure to the low melting (m.p. 126.degree.) photodimer (I) and the .alpha.-truxillic structure to the high melting (m.p. 226.degree.) photodimer (II). These results modify previous reports which assigned the .delta.-truxinic structure to the low melting isomer. The conformational properties of these mols. were investigated by comparing the exptl. dipole moments with contour maps of calcd. dipole moments as a function of the internal rotation angles, and with conformational energy maps. These structurally crowded mols. experience drastic restrictions of the conformational space available, so that they exist in well defined, thermodynamically preferred conformations.
- L10 ANSWER 34 OF 36 CA COPYRIGHT 2003 ACS
- AN 72:54946 CA
- TI .beta.- and .delta.-Truxinic acids
- AU Freedman, M.; Mohadger, Y.; Rennert, J.; Soloway, S.; Waltcher, I.
- CS City Coll., City Univ. of New York, New York, NY, USA
- SO Organic Preparations and Procedures (1969), 1(4), 267-9 CODEN: OGPPAC; ISSN: 0375-9512
- DT Journal
- LA English
- AB HO(CH2)3OH (38 g), 148 g trans-PhCH:CHCO2H and 10 g p-MeC6H4SO3H refluxed 3 hr in 150 ml PhMe yielded 85% (PhCH:CHCO2CH2)2CH2 (I), m. 87-8.degree. (MeOH), with characteristic ir spectrum. I (3 g) refluxed 1 week in 300 ml C6H12 under irradn. from a 275 W-sun-lamp and the soln. cooled gave 0.2 g bridged cyclobutane diester (II), which was sapond. and acidified to yield .delta.-truxinic acid quant. The C6H12 filtrate evapd. and the oily residue triturated with 10 ml 1:1 Et2O-petroleum ether gave 1.7 g II isomer (III), C21H2OO4, m. 162-3.degree., characterized by ir spectrum. Sapon. of III gave a quant. yield of .beta.-truxinic acid. The overall yield of II and III was 63%.
- L10 ANSWER 35 OF 36 CA COPYRIGHT 2003 ACS
- AN 69:86735 CA
- TI Acetylcholine. XII. 3,4-Diphenylthiophene-2.5-dicarboxylic acid bis [(.beta.-diethylamino)ethyl ester methiodide], a curarelike muscle-relaxant ester
- AU Dann, O.; Bamberg, K. J.; Sucker, H.
- CS Univ. Erlangen-Nuernberg, Erlangen-Nuernberg, Fed. Rep. Ger.
- SO Pharmazie (1968), 23(3), 135-45 CODEN: PHARAT; ISSN: 0031-7144
- DT Journal

LA German

GI

AΒ

For diagram(s), see printed CA Issue.

The muscle-relaxing properties of quaternized amino alc. esters of 3,4-diphenyl- (I), 3,4-dimethyl- (II), 3,4-di(2-furyl)- (III), and 3,4-bis(5-nitro-2-furyl)thiophene-2,5-dicarboxylic acid (IV); phenanthreno[9,10-c]thiophene-1,3-dicarboxylic acid (V); and 2,3-diphenylbenzene-1,4- (VI), 3,6-diphenylbenzene-1,2- (VII), and 2,5-diphenylfuran-3,4-dicarboxylic acid (VIII) were detd. I (10 g.) was boiled with 300 ml. SOC12 and worked up to give 8.1 g. I dichloride (IX), m. 123-4.degree.. Similarly prepd. were 37% II dichloride (X), m. 67-73.degree:; III dichloride (XI), 91%, m. 90.5-1.5.degree. (ligroine); IV dichloride, m. 92.5-95.degree. (C6H6); V dichloride (XII), 37%, m. 193-4.degree. (C6H6); and VI dichloride (XIII), 80%, m. 153.5-56.degree. (decompn.) (ligroine). VIII (5.5 g.) was added in small portions with stirring to an ice-cold suspension of 16 g. PCl5 in 55 ml. Et2O, stirred 30 min., and worked up to give 4.8 g. VIII dichloride (XIV), m. 120-1.degree. (twice from ligroine). Crude II in dioxane was treated with CH2N2 in Et2O, kept 3 hrs., and worked up to give 36% di-Me ester, m. 171.5-2.5.degree. (also prepd. by heating X and MeOH), which was refluxed in methanolic KOH and worked up to give pure II, decomp. 324-7.degree.. Similarly, III (at -5.degree.), gave 90% di-Me ester, m. 129.degree. (twice from AcOH), which, at -5.degree. in Ac2O, was nitrated with HNO3 (d. 1.52), stirred 1 hr., and worked up to give IV di-Me ester, m. 182-4.degree., which refluxed 2 min. in methanolic KOH and worked up gave IV, m. 258.degree. (decompn.). A suspension of 2 g. 1,4-dimethyl-2,3diphenylbenzene in 60 ml. C5H5N and 20 ml. H2O contg. 25.3 g. KMnO4 was refluxed 2 hrs. and worked up to give 2.2 g. VI, m. 308-11.degree.. IX (15.6 g.) and 25.2 g. .beta.-diethylaminoethanol (DEAE) was refluxed 6 hrs. in 500 ml. dry C6H6 and worked up to give 14.8 g. I bis(.beta.-diethylaminoethyl ester), m. 76.5-77.degree. (ligroine); dipicrate m. 175.5-77.degree. (1:1 Me2CO-H2O); di-HBr salt m. 185.5-6.5.degree. (Me2CO-iso-PrOH); dimethiodide m. 212-13.degree. (decompn.); bis(benzyl bromide) decompd. 191.degree., m. 240-7.degree. (EtOH-EtOAc). The following were prepd. II bis (.beta.-diethylaminoethyl ester), 70% [di-HBr salt m. 212.5-14.degree. (decompn.); dimethiodide m. 202.5-2.5.degree. (decompn.)]; III bis(.beta.-diethylaminoethyl ester), 61%, n22D 1.459, by shaking XI and DEAE in C6H6 66 hrs. at room temp. and working up [di-HBr salt, m. 179.5-81.degree. (decompn.); dimethiodide m. 177.5-79.degree. (decompn.)]; IV bis(.beta.-diethylaminoethyl ester), 47%, m. 42-7.degree. (dimethiodide m. 192-5.degree.); VI bis(.beta.diethylaminoethyl ester), 62%, n22D 1.540 [di-HBr salt m. 185.5-7.5.degree. (EtOAc:EtOH); dimethiodide, m. 234-5.degree. (decompn.)]; VIII bis(.beta.-diethylaminoethyl ester), 74% [di-HBr salt m. 180-1.degree. (3:1 Me2CO-EtOH); dimethiodide m. 185.5-87.degree. (decompn.)] I bis (.beta.-dimethylaminoethyl ester) 65%, m. 69-79.degree. (ligroine) [dimethiodide decompd. 225-50.degree. (EtOH)]; and V bis (.beta.-diethylaminoethyl ester), 90%, [di-HCl salt, decomp. 211-12.5.degree.; dimethiodide m. 215-16.degree. (decompn.)]. XII (1.25 g.) and 1.1 g. MeOH refluxed in 5 ml. C6H6 and cooled pptd. 0.85 g. V di-Me ester, m. 118-19.degree.. DEAE (4.7 g.) in 50 ml. Me2CO was added to 12 g. VII anhydride suspended in 250 ml. dry refluxing Me2CO, and the mixt. refluxed 20 min. to ppt. 14.6 g. of the half ester, m. 205-22.degree., difficulty sol. in 2N NaOH and 2N HCl. This intermediate (8.35 g.) and 5.4 g. .beta.-diethylaminoethyl chloride was refluxed 6.5 hrs. in 160 ml. dry iso-PrOH and worked up to give 7.3 g. VII bis (.beta.-diethylaminoethyl ester), m. 99-100.degree. (ligroine and petroleum ether); di-HBr salt m. 193-5.degree.; dimethiodide m. 206.5-7.5.degree. (decompn.). The anhydride (5 g.) of cis, cis, cis, cis-3,6-diphenyl-1,2,3,6-tetrahydrobenzene-1,2-dicarboxylic acid in 80 ml. HCONMe2 was hydrogenated at atm. pressure and room temp. over Pd(OH)2 on BaSO4 and worked up to give 3.3 g. anhydride of cis, cis, cis, cis-3,6-diphenylcyclohexane-1,2-dicarboxylic acid, m. 220-2.degree. (EtOAc). A soln. of 4.8 g. 2,7-diaminodiphenylene sulfone and 14 g. di-Et diacetylsuccinate in 20 ml. AcOH was refluxed 45 min. and cooled to ppt. 12.5 g. XV, m. 251-3.degree. (BuOH:AcOH), sapond. to the free acid by methanolic KOH. A mixt. of 1.28 g. 2,2'-dihydroxy-5,5'dimethyldeoxybenzoin in 2N NaOH and 1 g. ClCH2CO2H soln. neutralized with

K2CO3 was refluxed 3 hrs. and worked up to give 2-hydroxy-2'-carboxymethoxy-5,5'-dimethyldeoxybenzoin, m. 159-61.degree. (60% EtOH), and 2,2'-dicarboxymethoxy-5,5'-dimethyldeoxybenzoin, m. 172-4.degree. (60% AcOH and 60% EtOH). Extensive biol. data are given.

L10 ANSWER 36 OF 36 CA COPYRIGHT 2003 ACS

AN 66:82172 CA

- TI Separation of trans-cinnamic, .alpha.-truxillic, and .beta.-truxinic acids by paper chromatography
- AU Sonntag, Friedrich I.
- CS IBM Watson Res. Center, Yorktown Heights, NY, USA
- SO Journal of Chromatography (1967), 26(2), 544 CODEN: JOCRAM; ISSN: 0021-9673
- DT Journal
- LA English
- AB The title acids were sepd. by chromatog. on Whatman No. 1 paper. EtOH-H2O-NH4OH (80:10:5) was used as solvent for a front migration of 30 cm. The sheets were dried, and sprayed with a mixt. of 300 mg. bromothymol blue and 0.25 ml. 30% aq. NaOH in 500 ml. EtOH. Rf values were cinnamic acid, 0.56; .alpha.-truxillic acid, 0.36; .beta.-truxinic acid, 0.27. Cl- had to be excluded from the mixt.

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